



Donostia International Physics Center

Lecture 2

Xenon

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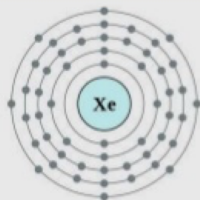
Xe Xe X E N O N Xe Xe Xe Xe

What is Xenon & its Physical Properties?

Xenon is a colourless, heavy and odourless noble gas; located in the periodic table in period five and group eight. Xenon occurs in the Earth's atmosphere in trace amounts and is generally unreactive. The boiling point of Xenon is 108.13°C (-162.5°F) and a melting point of -111.8°C . The atomic number for Xenon is 54 and its atomic weight is 131.293. Its electron configuration is 2,8,18,18,8.

The Discovery of Xenon

Scottish chemist William Ramsay and English Chemist Morris Travers discovered Xenon on the 12th of July, 1898. Both chemists found Xenon in the residue left over from evaporating components of liquid air. Xenon is a Greek word meaning "Foreign", "Strange" and "Guest".



Xe Xe X E N O N Xe Xe Xe Xe

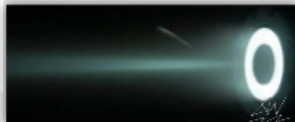
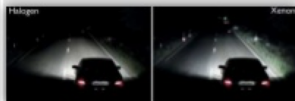
Chemical Properties

Xenon is a unreactive gas that is regarded as completely inert until Neil Barlett, in 1962 reported the synthesis of Xenon Haxafluoroplatinate. In a gas filled tube xenon emits blue light when excited by electrical discharge.

Uses

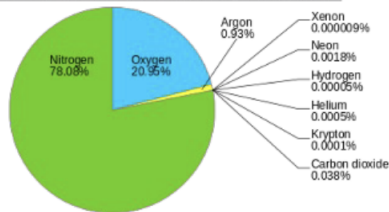
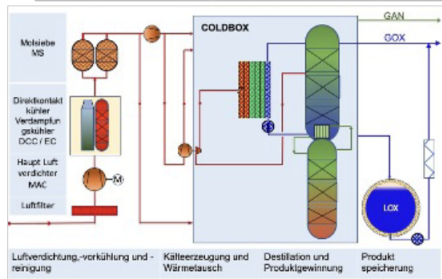
Xenon is used in:

- ❖ photographic flash lamps
- ❖ Stroboscopic lamps
- ❖ High-intensitive arc-lamps for motion picture projection
- ❖ High-pressure arc lamps to product ultraviolet light (solar simulators).
- ❖ Xenon 'blue' headlights and fog lights are used on some vehicles and are said to be less tiring on the eyes. They illuminate road signs and markings better than conventional lights.



Xenon Production

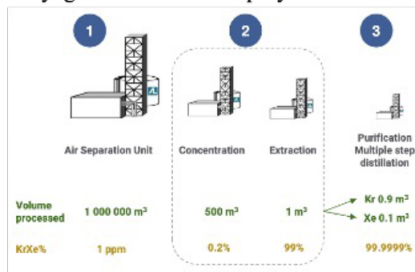
→ xenon is a by-product of air liquification predominantly for steel production



→ cryogenic distillation: liquify & boil-off

LOX: increased fraction of Xe, Kr, ...

- second round of distillation
 - further xenon-krypton enhanced 'crude'
 - further distilled to separate Xe and Kr
 - another distillation for highest purity Xe (6.0 or higher)
- Nomenclature: $5.0 = 10^{-5} = 99.999$ purity



Xenon Production Bottlenecks

There exist O(500) (sizable) air liquification plants all over the world

Only a small fraction ($\sim 1/5$) is equipped to produce rare gases

- traditionally high fraction in the former Soviet Union = Russia, Ukraine,...
- China growing rapidly...

Crude Xe/Kr production and Xe/Kr purification often in different countries

Adding Xe/Kr production requires to specially equip ASU's:

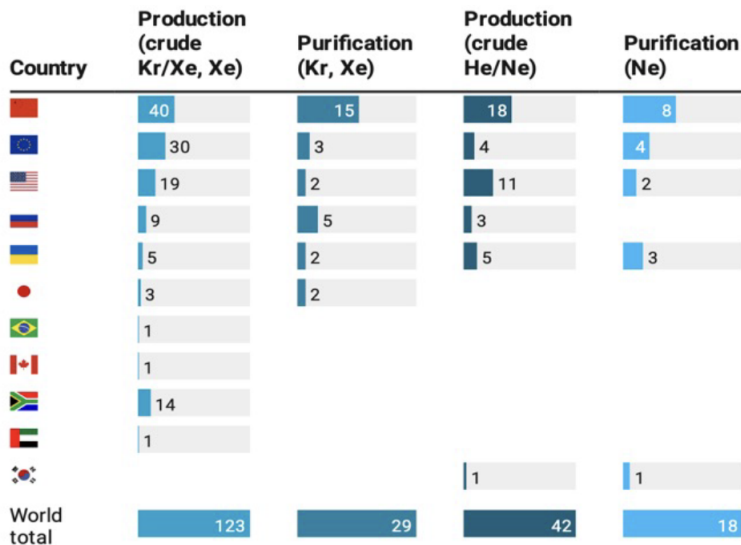
- financially not worth while to modify existing plants (cost/efficiency)
- financial threshold: O(80M€)
- newly built ASU's or added upon replacement of existing ASU's
 - if prices are expected high enough for long enough \longleftrightarrow return of invest

Overall world-wide xenon production (before Ukraine war):

\sim linearly growing. Now ca. 65 t / year; growth ca. 2t/year

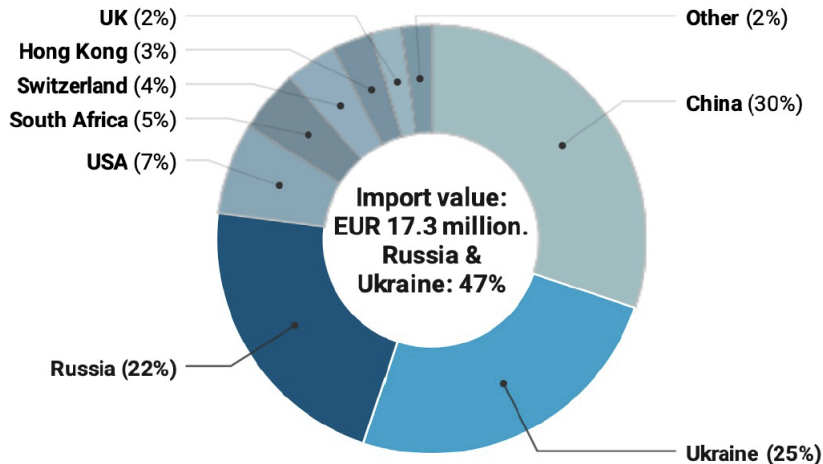
\sim ca. 90% are fixed term contracts with industry \rightarrow small spot market few t/year

Xenon Production by Country



Xenon from Russia and Ukraine

EU import value of rare gases¹⁶ by origin, in 2021



Source: Eurostat Comext (2022).

It's the economy stupid

- Xe/Kr is a by-product of steel production – otherwise extremely expensive
↔ connected to the steel market ↔ overall economic development
- Who pays for Kr/Xe separation
↔ Kr and Xe markets: demand ↔ uses
- The Krypton market:
Insulation windows ↔ energy costs for heating
lighting ↔ transition to LED's
propellant for thrusters ↔ Xe price
- Typical Xe price range: 5-20\$ /liter ↔ threshold for investing into new ASU's
- Past price excursions due to unexpected events:
 - Iridium satellite plans
 - plasma displays
 - now: Ukraine war...

A Volatile Market

Recent developments during due to Ukraine war:

- a large fraction of the Ukrainian production is missing
- embargo policies \leftrightarrow Russian suppliers

Consequences:

- very volatile market
- consumers moving to alternative solutions:
 - e.g. Starlink → argon (lifetime of thrusters \leftrightarrow cost of replacement)
 - ...
- in addition: general market trends
- prices went very high even beyond 1000\$/liter

Changes:

- A) gas companies think about new ASU's in other countries
 - time scales?
 - is on-going...
- B) embargo by-passing: xenon from Kasachstan? ...more from China?, ...
 - prices have significantly dropped, even \$10/L , market still volatile...

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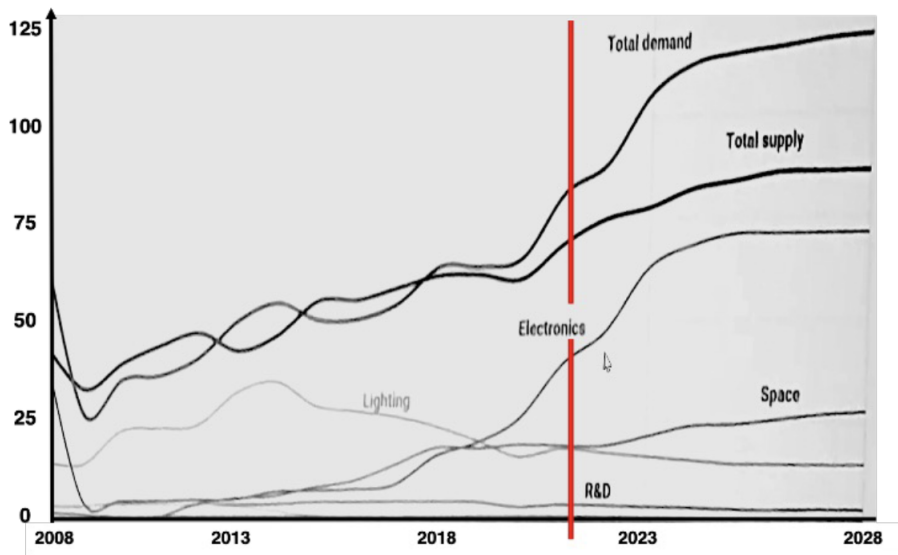
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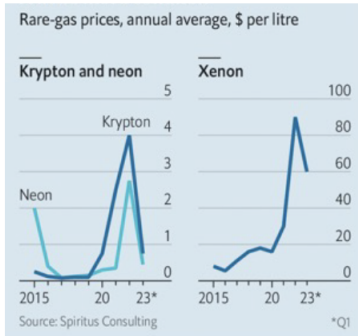
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Supply versus Demand



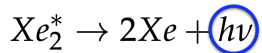
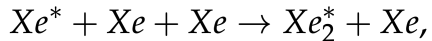
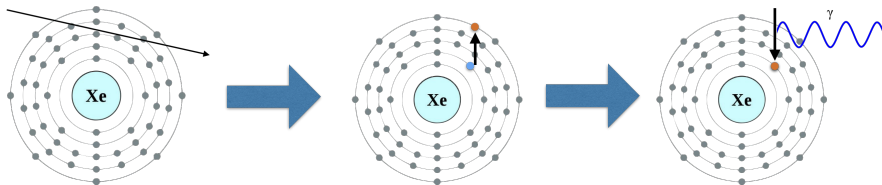
Xenon Price



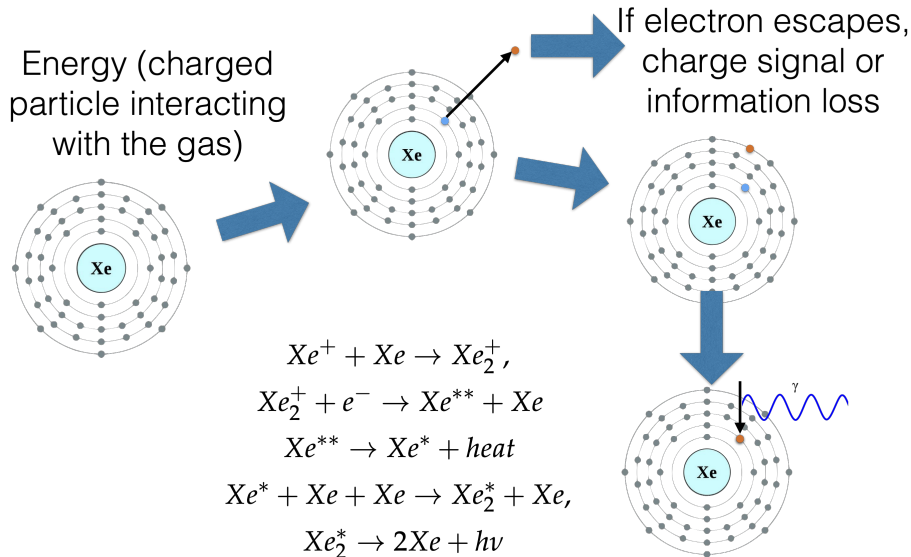
The Economist

Ionisation and scintillation in Xe

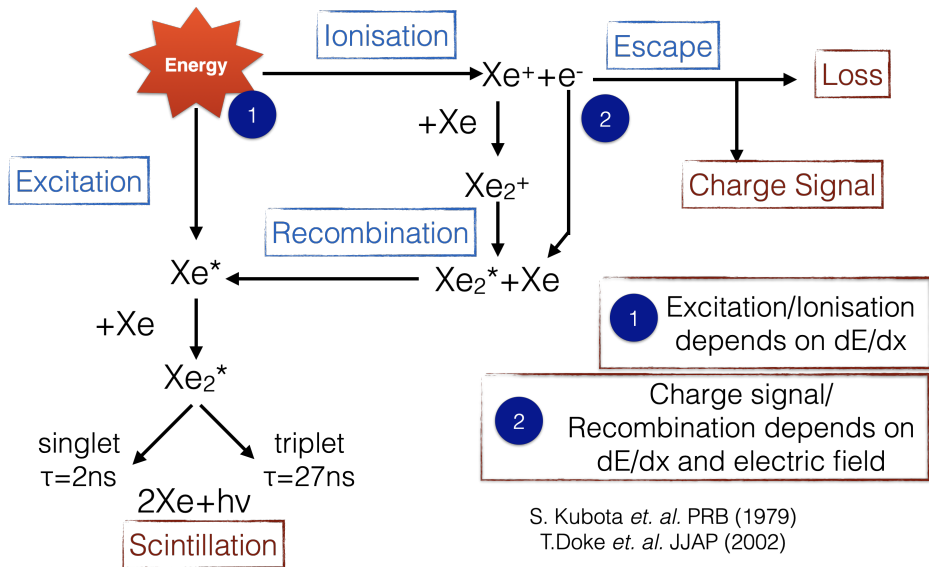
Energy (charged particle interacting with the gas)



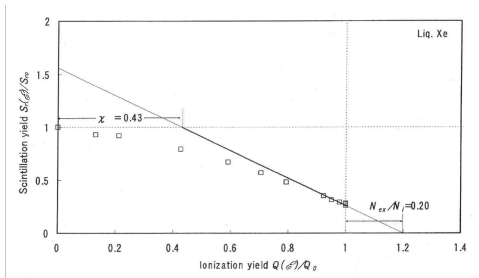
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Ionisation and scintillation in Xe

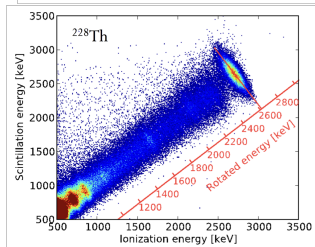


Anticorrelation between ionisation and scintillation in Xe

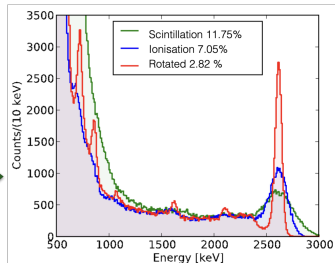


Perfect anti-correlation
of the charge and light signal

Doke, T. et al., 2002, Jpn. J. Appl. Phys

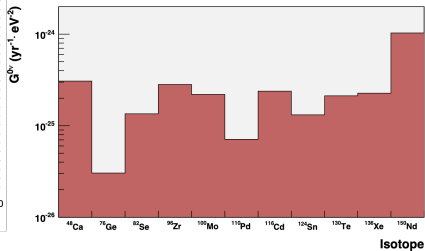
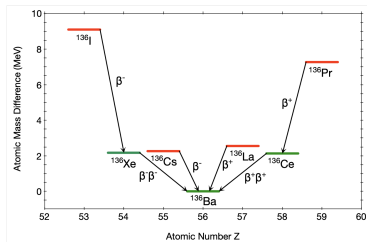


LXe experiments
use this anti-correlation to
get a better energy
measurement

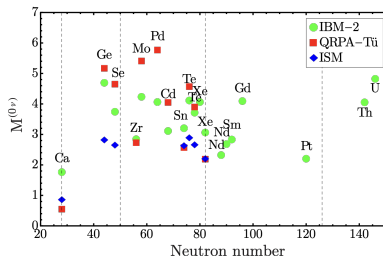


^{136}Xe is a good isotope for $\beta\beta 0\nu$ searches

X

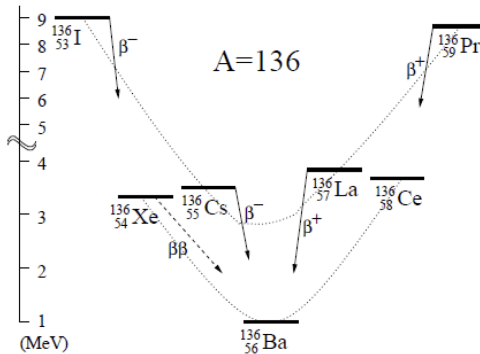


Relative atomic mass	^{136}Xe	135.907219(8) [24]
Q value $^{136}\text{Xe} \rightarrow ^{136}\text{Ba}$		2457.83(37) keV [25] 2458.7(6) keV [26] 2458.1(3) keV (average)
$G^{0\nu}$ ($10^{-15} \text{ year}^{-1}$)		14.58 [27] 14.54 [28]
$0\nu\beta\beta$ decay NME		2.19 (ISM) [29] 3.05 (IBM-2) [30] 2.46 (QRPA) [31] 2.91 (QRPA) [32] 4.12 (EDF) [33] 4.32 (EDF) [34]



^{136}Xe is a good isotope for $\beta\beta 0\nu$ searches

- Xenon is a good candidate for $\beta\beta 0\nu$ search
 - Q-value larger than energy of gammas from most natural radionuclides
 - Relatively easy to enrich in Xe-136 isotope
 - (no chemistry, centrifuge eff $\sim dm=4.7$ a.m.u.)
 - No need to grow high-purity crystals, continuous purification is possible (and relatively easy for a noble gas), more easily scalable
 - No long-lived cosmogenically activated isotopes
 - Final state (Ba-136 ion) can, in principle, be tagged, greatly reducing backgrounds



Exercise 1

- How much money would you need to buy the raw material for your ultimate $\beta\beta 0\nu$ experiment in 2017? And in 2023? Assume you will need O(100) tons of Xe.